to close the gap and explain the regiochemical and stereochemical results, the theory of such interactions, particularly with respect to 1,3-dipolar cycloadditions, has not been adequately developed to allow one to reach any firm conclusions at this time. Other factors including those discussed above may also play a significant role.

The main thrust of the work that has been described in this Account was experimental. The results that have been presented on the regio- and stereochemical outcome of cycloadditions of difluoroallene and fluoroallene will be pertinent to any further theoretical rationale that is presented to explain the effect of substituents on this broad class of reactions. Moreover it should be evident that appropriately substituted allenes, with their perfectly aligned allylic bonds, constitute ideal substrates for use in testing the theories of π -facial selectivity.

Sponge Phospholipids[†]

CARL DJERASSI* and WAI-KWAN LAM[‡]

Department of Chemistry, Stanford University, Stanford, California 94305 Received September 5, 1990 (Revised Manuscript Received December 19, 1990)

Sponges are simple cell aggregates that have thrived since the Cambrian Age in virtually all aquatic environments: marine and freshwater, from the equator to the poles, from shallow waters to enormous depths.¹ What is special about these ancient cells and their boundaries? What adaptive biochemical features exist in these oldest of the metazoans?

At a time when cell membrane biochemistry is gaining widespread attention, the discovery of well over 100 unprecedented sterols in sponges²⁻⁵ followed by over 50 phospholipids⁶ with unique structures and biosynthetic origins seems to have mystified further the riddle of membrane sterol-phospholipid interaction.⁷⁻¹⁴ Are these sponge lipids indeed cell membrane components? What is their functional significance? In this Account we will attempt to unify the diverse findings on sponge phospholipids of the diacylglycerol type from comparative and evolutionary perspectives. Ether phospholipids, such as the plasmalogens, will not be considered.

Background

Our interest in sponge phospholipids is traceable to the earlier research on marine sterols.²⁻⁵ In contrast to the monotony of terrestrial sterols (Figure 1), marine sterols show complex variations in both the tetracyclic nucleus and the side chain (Figures 2 and 3). The sterol mixtures of some sponges are composed entirely of the unique A-nor and 19-nor nuclei;3 in others the isooctyl side chain of cholesterol is extended by alkyl branches and three-membered rings at various positions (Figure 3).3-5 These structural oddities have attracted intensive biosynthetic studies.² The deliberate production of special lipids has, however, made even more puzzling their biological role in the organisms.

Sponges typically contain mixtures of the 24-alkyl sterols found in plants and cholesterol. However, a

Carl Djerassi is Professor of Chemistry at Stanford University. His chemical autobiography, Steroids Made it Possible, appeared as Vol. 4 in the series Profiles, Pathways, and Dreams, J. I. Seeman, Editor (American Chemical Society; Washington, DC, 1990) and contains more information than any casual reader may possibly desire

Wai-kwan Lam received his B.Sc. and M.Sc. degrees from the University of Hong Kong and the University of London, respectively. He had nearly completed the requirements for his Ph.D. at Stanford University when he met an untimely death

significant number of sponges possess substantial amounts of novel sterols, which prompted our early speculation^{5a} that they were cell membrane components. Cholesterol is a membrane stabilizer:8-13 it maintains the integrity of animal cell membranes by reducing the effective area of phospholipids (known as the condensing effect) and adjusting the flexibility and permeability of the latter. To achieve these ends the sterol molecule in general should have a rigid planar nucleus, a 3β -hydroxy group, C-18 and C-19 angular methyls, 17β and 20R configurations, and a C_6 – C_{10} side chain (Figure 1).^{5,8,9,14} Some sponge sterols contradict this rule (Figures 2 and 3):5,9 the A-nor stanols have no 3β -hydroxyl function; the 19-nor stanols have no methyl group at C-19; and sterols with extended side chains interact poorly with the phospholipids in mam-

- †Phospholipid Studies of Marine Organisms. 25. Part 24: Zimmerman, M.; Hoberg, M.; Ayanoglu, E.; Djerassi, C. Lipids 1990, 25, 383–390.

 Deceased, September 14, 1990.
- (1) Bergquist, P. R. Sponges; University of California Press: Berkeley,
- (2) Djerassi, C. In Roussel Award Lecture, Paris, 1988; pp 17-50. Djerassi, C. Steroids Made It Possible; American Chemical Society: Washington, DC, 1990; pp 114-127. Giner, J.-L.; Djerassi, C. Tetrahedron Lett. 1990, 31, 5421-5424. Garson, M. J. Nat. Prod. Rep. 1989, 6, 143-170.
- (3) Minale, L.; Sodano, S. In Marine Natural Products Chemistry; Faulkner, D. J., Fenical, W. H., Eds.; Plenum Press: New York, 1977; pp 87-109.
- (4) Djerassi, C. Pure Appl. Chem. 1981, 53, 873-890.
- (5) (a) Carlson, R. M. K.; Tarchini, C.; Djerassi, C. In (IUPAC) Frontiers of Bioorganic Chemistry and Molecular Biology; Ananchenko, S. N., Ed.; Pergamon Press: Oxford, 1980; pp 211-224. (b) Ikekawa, N. In Sterols and Bile Acids; Danielson, H., Sjovall, V., Eds.; Elsevier:
- Amsterdam, 1985; pp 199-230.

 (6) Lam, W.-k.; Hahn, S.; Ayanoglu, E.; Djerassi, C. J. Org. Chem. 1989, 54, 3428-3432 and preceding papers in the series Phospholipid Studies of Marine Organisms.
- (7) Deinum, G.; van Langen, H.; van Ginkel, G.; Levine, Y. K. Biochemistry 1988, 27, 852-860.
 (8) Yeagle, P. L. Biochim. Biophys. Acta 1985, 822, 267-287.
 (9) Bloch, K. E. Crit. Rev. Biochem. 1983, 14, 47-92.
- (10) Hui, S. W. In Biology of Cholesterol; Yeagle, P. L., Ed.; CRC
- Press: Boca Raton, FL, 1988; pp 213-231.
 (11) Huang, C. H. Lipids 1977, 12, 348-356.
 (12) Jain, M. Introduction to Biological Membranes, 2nd ed.; Wiley:
- New York, 1988.
- (13) Vance, D. E., Vance, J. E., Eds. Biochemistry of Lipids and Membranes; Benjamin/Cummings: Menlo Park, CA, 1985. (14) Craig, I. F.; Boyd, G. S.; Suckling, K. E. Biochim. Biophys. Acta 1978, 508, 418-421.

Figure 1. Generalized terrestrial sterol structure.

stigmastero

(plants)

Figure 2. Generalized sponge sterol nuclei.

malian type model membrane systems.¹⁴ If complex side chains are located in the hydrophobic region of the phospholipid bilayer, then do the phospholipids of sponges, in particular the fatty acyl components, display complementary structural novelties?

Chemical Diversity of Sponge Phospholipids

A. The Demospongic Acids. Studies over the past two decades have yielded surprising results. Litchfield and collaborators reported 15-20 high levels (34-79% of total) of C₂₄-C₃₄ fatty acids in numerous sponge species as noted initially by Bergman and Swift. These exceptionally long chain fatty acids, mostly polyunsaturated, were dubbed "demospongic acids".20 Of special interest is the 5Z,9Z diene pattern of the many $\mathrm{C}_{24}\text{--}\mathrm{C}_{28}$ acids studied (e.g., 1 and 2 in Figure 4). These molecules contrast sharply with their familiar C₁₄-C₂₂ counterparts where the methylene-interrupted unsaturation pattern is common (e.g., $\Delta^{9,12,15}$ -18:3, linolenic acid).¹³ The demospongic acids, notably, were found mostly in the amino phospholipids phosphatidylethanolamines (PE) and phosphatidylserines (PS), 18 whereas the short-chain fatty acids predominated in the common "animal phospholipids", 13 the phosphatidylcholines (PC).

Our venture into the phospholipid field began with the marine sponge Aplysina fistularis,22 whose principal sterols are aplysterol and its 25,26-dehydro analogue

(15) Litchfield, C.; Tyszkiewitz, J.; Dato, V. Lipids 1980, 15, 200-202. (16) Litchfield, C.; Tyszkiewitz, J.; Marcantonio, E. E.; Noto, G. Lipids (Figure 3). The phospholipid acyl components of this sponge comprise 85% conventional C₁₄-C₂₀ acids (some branched, as is common among bacteria) and 15% C₂₇₋C₃₀ demospongic types. The latter, again mainly in PE and PS, all possess the unusual 5Z,9Z diene system and have an additional methyl branch or double bond at the ω 7 carbon (e.g., 3 in Figure 4).

Molecular models of one of the major demospongic acids in A. fistularis, 22-methyl-5,9-octacosadienoic acid $(22\text{Me-}\Delta^{5,9}-28:2)$, suggested²² that if the 5Z,9Z double bonds were stacked upon each other in the cell membrane to achieve maximum π overlap, the 22-methyl branch would sterically interfere with the terminal methyl groups of the sterol side chain. In yeast, a disordering of the phospholipids is caused by the C-24 methyl group of ergosterol.^{8,9} Apparently this disordering effect plays a compensatory role in fluidizing the cell membrane of yeast and some plants where highly unsaturated fatty acids are lacking.9 In the case of sponges the accumulation of C₂₄-C₃₄ acyl chains in the membrane would reduce its flexibility and render membrane-associated reactions more difficult than with their standard C₁₄-C₂₂ counterparts. So is the steric interference a means to restore membrane fluidity in the sponge? Or is the sponge membrane distinct in structure and function as Litchfield and Morales speculated?¹⁸

Such conjectures prompted us to undertake a systematic investigation of the phospholipid fatty acids of various sponges. Some of the remarkable features of demospongic acids (Figure 4) include the presence of cyclopropyl, 25 α -acetoxy, 26 or α -methoxy 27,28 functions (of R configuration) in C_{22} – C_{30} acyl chains; methyl branches at various positions;^{28–31} the unique 6-bromo- $\Delta^{5,9}$ moiety of C_{26} – C_{27} acids with or without methyl branching;^{6,32} and the abundance of numerous odd-chain (C_{21} – C_{27}) analogues.³³ Common among the unsaturated acids is again the 57.07 diamenet unsaturated acids is again the 5Z,9Z diene pattern, although some long-chain monoenes have been identified (e.g., 19-Me,24-Me- Δ^5 -25:1, 8 in Figure 4).²⁹ The Δ^5 double bond, incidentally, is unusual and exists mainly in the C_{16} – C_{18} fatty acids of marine invertebrates^{29,34–36} and some plants.^{37–39} Recently the structural range of demospongic acids has been widened further by Carballeira and co-workers, who observed, in addition to the above structural features, 40-42 the

¹⁹⁷⁹, 14, 619-622. (17) Litchfield, C.; Marcantonio, E. E. Lipids 1978, 13, 199-202.

⁽¹⁸⁾ Morales, R. W.; Litchfield, C. Biochim. Biophys. Acta 1976, 431, 206-216

⁽¹⁹⁾ Jefferts, E.; Morales, R. W.; Litchfield, C. Lipids 1974, 9, 244-247. (20) Litchfield, C. In Aspects of Marine Biology; Harrison, R. W. Cowden, R. R., Eds.; Academic Press: New York, 1976; pp 183-200. (21) Bergman, W.; Swift, A. N. J. Org. Chem. 1951, 16, 1206-1221. (22) Walkup, R. D.; Jamieson, G. C.; Ratcliff, M. R.; Djerassi, C. Lipids

^{1981, 631-646.}

⁽²³⁾ Ayanoglu, E.; Wegmann, A.; Pilet, O.; Marbury, G. D.; Hass, J. R.; Djerassi, C. J. Am. Chem. Soc. 1984, 106, 5246-5251 and references cited therein.

⁽²⁴⁾ Tomer, K. B.; Jensen, N. J.; Gross, M. L. Anal. Chem. 1986, 58, 2429-2433

⁽²⁵⁾ Lankelma, J.; Ayanoglu, E.; Djerassi, C. Lipids 1983, 18, 853-858. (26) Ayanoglu, E.; Kurtz, K.; Kornprobst, J. M.; Djerassi, C. Lipids 1985, 20, 141-144.

⁽²⁷⁾ Ayanoglu, E.; Kornprobst, J. M.; Aboud-Bichara, A.; Djerassi, C. Tetrahedron Lett. 1983, 24, 1111-1114.

⁽²⁸⁾ Ayanoglu, E.; Popov, S.; Kornprobst, J. M.; Aboud-Bichara, A.; Djerassi, C. *Lipids* 1983, 18, 830–836.

 ⁽²⁹⁾ Dasgupta, E.; Ayanoglu, E.; Djerassi, C. Lipids 1984, 19, 768-776.
 (30) Ayanoglu, E.; Walkup, R. D.; Sica, D.; Djerassi, C. Lipids 1982,

⁽³¹⁾ Lawson, M. P.; Bergquist, P.; Cambie, R. C. Biochem. Syst. Ecol. 1984, 12, 375-393

⁽³²⁾ Wijekoon, W. M. D.; Ayanoglu, E.; Djerassi, C. Tetrahedron Lett. 1984, 3285-3288

⁽³³⁾ Bergquist, P.; Lawson, M. P.; Lavis, A.; Cambie, R. C. Biochem.

<sup>Syst. Ecol. 1984, 12, 63-84.
(34) Carballeira, N. M.; Maldonado, L. Lipids 1986, 21, 470-471.
(35) Carballeira, N. M.; Maldonado, L.; Porras, B. Lipids 1987, 22,</sup>

⁽³⁶⁾ Gillan, F. T.; Stoilov, I. L.; Thompson, J. E.; Hogg, R. W.; Wilkinson, C. R.; Djerassi, C. Lipids 1988, 23, 1139-1145.
(37) Takagi, T.; Itabashi, Y.; Kaneniwa, M.; Mizukami, M. Yukagaku

^{1983, 32, 367-374} and references cited therein.

⁽³⁸⁾ Madrigal, R. V.; Smith, C. R., Jr. Lipids 1975, 10, 502-504. (39) Davidoff, F.; Korn, E. D. J. Biol. Chem. 1963, 238, 3199-3209.

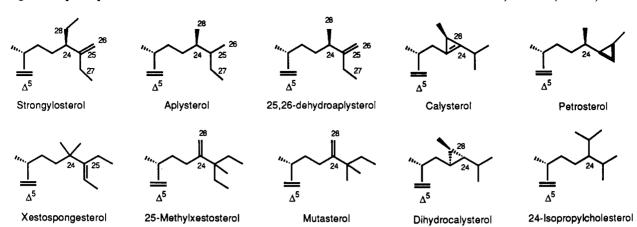
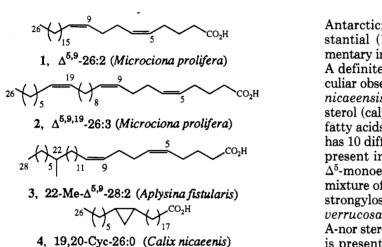


Figure 3. Examples of sponge sterol side chains.



5, 2-OAc-25:0 (Polymastia gleneni)

6, 2-OMe- Δ^{21} -28:1 (Higginsia tethyoides)

7, 6-Br-Δ^{5,9}-26:2 (Hymeniacidonid Sponge)

8, 19-Me,24-Me-Δ⁵-25:1 (Strongylophora durissima)

Figure 4. Some sponge phospholipid fatty acids (demospongic acids).

 α -hydroxy function^{43,44} and the rare Δ ^{6,11}-diene system.⁴⁵ It is noteworthy that all demospongic acids identified so far predominate in the PE and PS fractions (ca. 75% vs 25% in PC).

Demospongic acids have been identified in marine and freshwater sponges collected from Siberia⁴⁶⁻⁴⁹ to the Antarctic;50 the total abundance in all cases is substantial (15-60%). Are demospongic acids complementary in structure and function to the sponge sterols? A definite answer is difficult to provide, but a few peculiar observations are worth noting. The sponge Calyx nicaeensis has three-membered rings in both its major sterol (calysterol, Figure 3) and one of its phospholipid fatty acids (4 in Figure 4). 25 Strongylophora durissima has 10 different methyl-branched long-chain fatty acids present in PE and PS, three of which are of the rare Δ^5 -monoenic type (e.g., 8 in Figure 4).²⁹ The sterol mixture of S. durissima contains mostly the novel sterol strongylosterol (Figure 3) and no cholesterol. Axinella verrucosa and Phakelia aruensis contain mixtures of A-nor sterols which coexist with the $\Delta^{5,9}$ -26:2 acid, which is present solely in PE and PS.51

The complementarity of novel sponge sterols and phospholipids is still uncertain. Although aplysterol (Figure 3) (60%) and the 22-Me- $\Delta^{5,9}$ -28:2 acid (6.1%) are present in significant amounts in A. fistularis, there is no evidence that they interact with each other in the cell membrane, since cholesterol (11%), the short-chain fatty acids (85%), and another unusual $\Delta^{5,9,23}$ -30:3 acid (5.6%) co-occur, most likely as membrane components.²² The sterols of Microciona prolifera are mostly (99%) "conventional", whereas over 30% of the total phospholipid acids are of the $\Delta^{5,9}$ -26:2 and $\Delta^{5,9,19}$ -26:3

B. Intact Sponge Phospholipids. A striking fea-

ture of sponges is the attachment of two identical demospongic acyl chains to the glycerol backbone, 52,53 in contrast to the conventional animal phospholipids which typically carry a saturated acyl group at C-1 of glycerol and an unsaturated one at C-2.13 The relative abundance of phospholipid classes varies with sponge species, but the amino phospholipids, especially PE, are

usually predominant.

⁽⁴⁰⁾ Carballeira, N. M.; Maldonado, M. E. Lipids 1989, 24, 371–374.
(41) Carballeira, N. M.; Shalabi, F.; Maldonado, M. E. Lipids 1989, 24, 235-237.

⁽⁴²⁾ Carballeira, N. M.; Reyes, E. D. Lipids 1990, 25, 69-71.
(43) Carballeira, N. M.; Lopez, M. R. Lipids 1989, 24, 89-91.
(44) Carballeira, N. M.; Shalabi, F.; Negron, V. Lipids 1989, 24, 229.
(45) Carballeira, N. M.; Maldonado, M. E. Lipids 1989, 24, 665-668.

⁽⁴⁶⁾ Dembitskii, V. M. Khim. Prir. Soedin. 1981, 4, 511-513; Chem.

Abstr. 1982, 96, 31906m. (47) Dembitskii, V. M. Khim. Prir. Soedin. 1981, 4, 513-515; Chem.

Abstr. 1982, 96, 17249t.
(48) Dembitskii, V. M. Zh. Evol. Biokhim. Fiziol. 1982, 18, 92-94;
Chem. Abstr. 1982, 96, 119403f.
(49) Dembitskii, V. M.; Chelomin, V. P. Izv. Akad. Nauk SSSR, Ser.

⁽⁴⁹⁾ Demoitskii, v. M.; Chelomin, v. P. 120. Arada. Natur SSSR, Ser. Biol. 1985, 1, 53-6; Chem. Abstr. 1985, 102, 129125y.
(50) Lam, W.-k.; Djerassi, C., unpublished observation.
(51) Ayanoglu, E.; Djerassi, C., unpublished observation.
(52) Dasgupta, A.; Ayanoglu, E.; Wegmann-Szente, A.; Tomer, K. B.; Djerassi, C. Chem. Phys. Lipids 1986, 41, 335-357.

⁽⁵³⁾ Dasgupta, A.; Ayanoglu, E.; Tomer, K. B.; Djerassi, C. Chem. Phys. Lipids 1986, 43, 101-111.

Scheme I Incorporation of 1-14C-Labeled Branched-Chain Fatty Acids into the Marine Sponge Jaspis stellifera

Isolation of the unstable demospongic molecular species^{52,53} and structure elucidation by fast atom bombardment and tandem mass spectrometry^{23,24,52,53} revealed the following unique PE molecular species: $\Delta^{5,9}$ -PE(26:2,26:2); $\Delta^{5,9,19}$ -PE(26:3,26:3); $\Delta^{5,9,21}$ -PE-(28:3,28:3); and $\Delta^{5,9,23}$ -PE(30:3,30:3). Cervonic ($\Delta^{4,7,10,13,16,19}$ -22:6) and arachidonic ($\Delta^{5,8,11,14}$ -20:4) acids, common among other organisms, exist in sponges as PC(22:6,22:6) and PC(20:4,20:4), respectively. Such "symmetric" phospholipids rarely exceed negligible quantities in higher organisms except for dipalmitoylphosphatidylcholine (DPPC), the mammalian lung surfactant.13

In addition to PE, PC, and PS, the other common head groups of sponge phospholipids are phosphatidylglycerol (PG)⁵¹ and phosphatidylinositol (PI).^{54,55} PG is typical of bacteria, and its high level in some sponges (usually associated with arachidonic acid, 20:4)⁵¹ suggests either the primitive nature of the latter or the incorporation of exogenous phospholipids. PI, however, plays a distinct role in sponge cell aggregation,54-58 the species-specific cell-recognition ability of sponges.

Demospongic Acid and Phospholipid **Biosynthesis**

There are several possible sources of sponge fatty acids: de novo biosynthesis, dietary intake, and incorporation from symbionts, with or without further modification. Morales and Litchfield demonstrated in M. prolifera that the most characteristic sponge fatty acids, the $\Delta^{5,9}$ -26:2 and $\Delta^{5,9,19}$ -26:3 acids (Figure 4), were biosynthesized via homologation of the respective short-chain precursors, palmitic (16:0) and palmitoleic $(\Delta^9-16:1)$ acids, followed by desaturation.⁵⁹ De novo biosynthesis starting from acetyl-CoA was found to be negligible.13

Our biosynthetic studies began with the branchedchain demospongic acids. By the incorporation of synthetic 1-14C-labeled 13- and 12-methylpentadecanoic acids (iso-15:0 and anteiso-15:0) into the Australian sponge Jaspis stellifera, it was shown⁶⁰ that the 25methyl-5,9-hexacosadienoic acid ($\Delta^{5,9}$ -iso-27:2) and its

(54) Weissmann, G.; Riesen, W.; Davidson, S.; Waite, M. Biochim.

Biophys. Acta 1988, 960, 351-364. (55) Zimmerman, M. P.; Hoberg, M.; Ayanoglu, E.; Djerassi, C. Lipids

1990, 25, 383-390.
(56) Muller, W. E. G.; Rottmann, M.; Diehl-Seifert, B.; Kurelec, B.; Uhlenbruck, G.; Schroder, H. C. J. Biol. Chem. 1987, 262, 9850-9858.
(57) Dunham, P.; Anderson, C.; Rich, A. M.; Weissmann, G. Proc. Natl. Acad. Sci. U.S.A. 1983, 80, 4756-4760.

(58) Burkart, W.; Jumblatt, J.; Simpson, T. L.; Burger, M. M., Biologie des spongiaires. *Colloq. Int. C.N.R.S.* 1979, *No.* 291, 239–246.
(59) Morales, R. W.; Litchfield, C. *Lipids* 1977, *12*, 570–576.
(60) Carballeira, N. M.; Thompson, J. E.; Ayanoglu, E.; Djerassi, C. J.

Org. Chem. 1986, 51, 2751-2756.

Scheme II Biosynthesis of Chiral Branched-Chain Demospongic Acids in the Marine Sponge Aplysina fistularis

Scheme III Biosynthesis of Straight-Chain Demospongic Acids in the Marine Sponge Microciona prolifera

12:0
$$+14:0$$
 16:0 26:0 $\Delta^{5.9.26:1}$ $\Delta^{5.9.26:2}$ $\Delta^{9.26:1}$ desaturation $\Delta^{9.16:1}$ $\Delta^{9.16:1}$ $\Delta^{19.26:1}$ $\Delta^{5.9.19.26:3}$ (exogenous)

24-methyl analogue ($\Delta^{5,9}$ -anteiso-27:2) in fact originate via chain elongation of the C₁₅ precursors (Scheme I). Our findings suggest that the short-chain precursors are obtained from bacteria¹² and then homologated and desaturated by the sponge.61

We investigated the biosynthesis of 22-methyl-5,9octacosadienoic acid (22-Me- $\Delta^{5,9}$ -28:2) in A. fistularis to determine the role of chirality at the branching site,61 using synthetic [1-14C]-dl-10-methylhexadecanoic acid (10-Me-16:0) and its 10R and 10S antipodes. Surprisingly, all three acids were incorporated and transformed in vivo into the d/l-22-Me- $\Delta^{5,9}$ -28:2 acid (Scheme II). No qualitative specificity between the two enantiomers was observed, although the natural acid is of the 22R-(-) configuration. The indiscriminate homologation of enantiomeric acyl precursors indicates that the chiral branching site, if located far enough from the carboxyl end, is not distinguished by the fatty acid synthase of the sponge. Judging from the above two biosynthetic studies, 60,61 the distal methyl branches in various demospongic acids are probably not introduced by the sponges themselves. Apparently fatty acid biosynthesis in sponges relies heavily on short-chain precursors originating from symbionts or diet: the rare 10-Me-16:0 acid, for instance, is present in the anaerobic marine sulfate-reducing bacteria which exist within A. fistularis.61

Elucidation of the complete biosynthetic pathway for straight-chain demospongic acids was our next task. For this purpose we selected the sponge M. prolifera, which Litchfield and co-workers used for their early biosynthetic studies.⁵⁹ Incorporation of various 1-¹⁴Clabeled fatty acids confirmed 62 that the predominant biosynthetic process for the C₂₆ acids proceeds via the homologation of short-chain precursors (Scheme III). The shortest saturated homologue amenable to chain elongation to the 26:0 acid was shown to be myristic (14:0) rather than the earlier assumed⁵⁹ palmitic acid (16:0). Lauric acid (12:0) was not homologated, 62 thus confirming the initial observation⁵⁹ of minimal de novo biosynthetic activity.

(61) Raederstorff, D.; Shu, A. Y. L.; Thompson, J. E.; Djerassi, C. J.

Org. Chem. 1987, 52, 2337–2346. (62) Hahn, S.; Stoilov, I.; Tam Ha, T. B.; Raederstorff, D.; Doss, G. A.; Li, H.; Djerassi, C. J. Am. Chem. Soc. 1988, 110, 8117-8124.

The most striking aspect of demospongic fatty acid biosynthesis is the desaturation process. Desaturation of the 26:0 acyl chain was found⁶² to commence at either the Δ^5 or Δ^9 loci, meaning that the second double bond may be inserted on either side of the existing one (Scheme III). This is contrary to the formation of polyunsaturated fatty acid formation in animals, 13 where the first double bond is normally introduced at the Δ^9 position, with the second and subsequent ones inserted between the first bond and the carboxyl terminus. Plants, however, do desaturate between the first double bond and the methyl end of the fatty acyl chain.13

The trienoic $\Delta^{5,9,19}$ -26:3 acid is formed in the marine sponge not from the saturated precursors, as in animal systems, ¹³ but rather from palmitoleic acid (Δ^9 -16:1), followed by desaturation at the Δ^5 or Δ^9 loci of the elongation product (Scheme III).⁶² This biosynthetic sequence indicates the lack of Δ^{19} -desaturase activity in the sponge, and more importantly it reflects the low affinity of the sponge's Δ^9 -desaturase enzyme for short-chain homologues like palmitic acid. Exogenous (bacterial?) palmitoleate (Δ^{9} -16:1) is used, instead, as the substrate for chain homologation (Scheme III).62

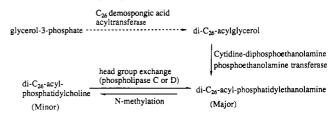
We extended our investigations to the freshwater sponges, which had been under limited study in the USSR.46-49 A series of radiotracer experiments demonstrated 63 that Δ^{19} -desaturase activity occurs in the freshwater sponge Ephydatia fluviatilis. As mentioned above, desaturation between the Δ^9 locus and the methyl terminus of the fatty acid is typical of plants. Animals, including marine sponges, normally do not desaturate beyond the Δ^9 position. On the basis of their fatty acid desaturation pathways, sponges^{62,63} possess both animal and plant characteristics.

The abundance of odd-chain demospongic acids in some sponge species raises the question of whether they share the same biosynthetic pattern with their evenchain counterparts. An affirmative answer was provided by our radiolabeling experiments with odd-chain precursors on a Puerto Rican sponge Xestospongia sp., 64 since its $\Delta^{5,9}$ -27:2 acid (>20% of total phospholipid acids) is formed mainly by homologation of the 15:0 or 17:0 acids followed by desaturation. De novo biosynthesis from propionate, which is known to be operative in higher organisms, 65 was found to be negligible.

By now a distinct pattern of demospongic acid biosynthesis has emerged. Sponges obviously possess an active homologating enzyme system acting on shortchain fatty acid precursors largely derived from exogenous sources (most likely bacteria or plankton). Fatty acid desaturation in sponges exhibits both animal and plant behavior. There is no conclusive evidence that sponges are incapable of de novo fatty acid biosynthesis, but so far the formation of short-chain acids from radiolabeled acetate has not been observed to any significant extent.

The mechanism of functionalization of sponge fatty acids is less well understood. Substituents like α -hydroxy, 43,44 α -methoxy, 27,28 α -acetoxy, 26 and bromo (C-6, vinylic)6,32 are apparently introduced after chain elon-

Scheme IV Pathways of Phospholipid Metabolism in the Marine Sponge Microciona prolifera



gation. The bromo function of the 6-Br- $\Delta^{5,9}$ -26:2 acid,⁶ for instance, has been shown in a Hymeniacidonid sponge to be introduced into the $\Delta^{5,9}$ -26:2 acid in the final step of biosynthesis.

The biosynthesis of intact sponge phospholipids proceeds⁶⁶ via the cytidine pathways as in other eukaryotes. The attachment of two identical acyl chains to PE appears, however, to be mediated by acyl- and phosphoethanolamine transferases specific for the demospongic acids (Scheme IV). The interconversion of phospholipid classes is appreciable, especially from PC to PE by head-group exchange and, conversely, by N-methylation. Unnatural PCs with demospongic acyl groups tend to be converted rapidly into the PE analogues after indiscriminate uptake by the sponge, thus highlighting the affinity of amino phospholipids for demospongic acids.

Cellular Studies

We recently demonstrated that demospongic acids (C₂₄-C₃₄, mostly unsaturated) and C₂₄-alkylated sterols are both found in the sponge membrane. 67-70 It must be stressed, however, that unconventional sterols are not universal among sponges. The cell membrane of Reniera sp., for instance, has been found to contain demospongic acids but only conventional type sterols.⁶⁹ Thus, demospongic acids do not always co-occur with novel sponge sterols. These findings point toward a highly intriguing biophysical relationship between unusual sponge membrane lipids, which we examined by using model membrane systems (vide infra).

Another objective of our cellular studies is to identify the biosynthetic sites of demospongic phospholipids and their distribution among different cell categories. Sponges are loose aggregates of a few cell types. 1,71 Storage cells are often present, such as the gray cells or spherulous cells in some sponge species. Most crucial to our sponge lipid studies, however, are the inter- or intracellular microbial symbionts,71 since they may contribute to the total lipid composition of the sponge or provide short-chain precursors for demospongic acid biosynthesis. Cell separation is essential, therefore, to fully characterize the unusual lipid molecules.

Different sponge cell types can be separated by the combined use of density gradients and fluorescence-

⁽⁶³⁾ Hahn, S.; Lam, W.-k.; Wu, I.; Silva, C. J.; Djerassi, C. J. Biol. Chem. 1989, 264, 21043-21046.
(64) Lam, W.-k.; Djerassi, C., unpublished observation.
(65) Horning, M. G.; Martin, D. B.; Karmen, A.; Vagelos, P. R. J. Biol.

Chem. 1961, 236, 669-672.

⁽⁶⁶⁾ Lam, W.-k.; Beatty, M. F.; Hahn, S.; Djerassi, C. Biochemistry

⁽⁶⁷⁾ Lawson, M. P.; Thompson, J. E.; Djerassi, C. Lipids 1988, 23, 1037-1048.

⁽⁶⁸⁾ Lawson, M. P.; Thompson, J. E.; Djerassi, C. Lipids 1988, 23, 741-749.

⁽⁶⁹⁾ Lawson, M. P.; Stoilov, I. L.; Thompson, J. E.; Djerassi, C. Lipids 1988, 23, 750-754

⁽⁷⁰⁾ Lawson, M. P.; Bergquist, P. R.; Cambie, R. C. Tissue Cell 1986, 18, 19-26.

⁽⁷¹⁾ Simpson, T. L. The Cell Biology of Sponges; Springer-Verlag: New York, 1984.

activated cell sorting. 55,58,72-74 As expected, an uneven distribution of demospongic acids and sterols among different cell types was observed. 55,72,73 The small surface cells are rich in demospongic acids but low in sterol content as compared to the internal, metabolically active cells. The relatively high levels of short-chain acids in the metabolically active cells is most likely due to the accumulation of symbionts and engulfed food particles, which may be the ultimate supplier of precursors for demospongic acid biosynthesis. No demospongic acid or unconventional sterol, however, has been found in the cultured microbial symbionts.⁷²

Radiotracer experiments with carefully separated archeocytes and choanocytes of M. prolifera revealed⁷³ fatty acid homologation occurring in both cell types. The elongation of short-chain precursors is the chief mode of demospongic acid biosynthesis in sponge cells, which require such acids for incorporation into the amino phospholipids. What is their membrane function?

Are Sponge Cell Membranes Unique?

The notion that the sponge cell membrane was unique¹⁸ arose primarily from the steric bulk and entropy anticipated from the long-chain cis-polyenoic phospholipids. Furthermore, the abundance of PE would contribute to membrane density due to H-bonding between the ethanolamine head groups. The absence of any visible anomaly under the electron microscope challenged this speculation.⁷⁵ The currently accepted "fluid mosaic model" of animal cell membranes is a trilaminate, bilayer assembly of phospholipids (5-15 nm) integrated with cholesterol and membrane proteins. 12,13,70 The sponge membrane fulfills such morphological criteria, 75 which lead to the erroneous argument⁷⁵ that demospongic acids and the unusual sterols were not membrane components.

Demospongic phospholipids are cell membrane components, although they coexist with either the conventional or novel sponge sterols⁶⁹ and cause no visible distortion of the bilayer.⁷⁵ In order to understand the interaction between demospongic acids and sterols, we employed model membrane studies, using both isolated sponge phospholipids and their synthetic analogues. These studies led to a surprising result: the unique sponge phospholipids do not interact with sterols in cell membranes!

In our first study, ⁷⁶ liposomes of either synthetic ⁷⁷ $\Delta^{5,9}$ -PE(26:2,26:2) or $\Delta^{5,9}$ -PC(26:2,26:2) underwent an endothermic phase transition at 42 °C (typical of DPPC),⁷⁸ which was unaffected by cholesterol (up to 20 mol %). Broadening of the phase transition was minimal. The usually higher phase-transition temperature of PE (due to H-bonding between head groups) was not observed, presumably because of the overriding effect of the long acyl chains. Replacing

(72) Zimmerman, M. P.; Thomas, F. C.; Thompson, J. E.; Djerassi, C.;
Streiner, H.; Evans, E.; Murphy, P. T. Lipids 1989, 24, 210-216.
(73) Lam, W.-k.; Djerassi, C., manuscript in preparation.
(74) Parks, D. R.; Herzenberg, L. A. Methods Enzymol. 1984, 108,

(75) Lethias, C.; Garrone, R.; Mazzorana, M. Tissue Cell 1983, 15,

(76) Ayanoglu, E.; Düzgünes, N.; Wijekoon, W. M. D.; Djerassi, C.

 Biochim. Biophys. Acta 1986, 863, 110-114.
 (77) Mena, P. L.; Djerassi, C. Chem. Phys. Lipids 1985, 37, 257-270.
 Mena, P. L.; Pilet, O.; Djerassi, C. J. Org. Chem. 1984 49, 3260-3264.
 (78) Mabrey, S.; Mateo, P. L.; Sturtevant, J. M. Biochemistry 1978, 17, 2464, 2469. 17, 2464-2468.

cholesterol with A-norcholesterol or petrosterol (Figure 3) gave similar results: the sterol was excluded from the phospholipid bilayer.

It is striking that these two membrane lipid types, being so abundant in various sponge species, should have no mutual interaction at all in the membrane. We speculated that the 5Z,9Z diene pattern and the exceptional chain length of the acyl groups were both responsible for the exclusion of sterols from demospongic phospholipids.

To examine the first factor, liposomes containing the synthetic phospholipid $\Delta^{6,9}$ -PE(26:2,26:2) or its PC analogue were prepared.⁷⁹ With unusual chain length but the conventional 6Z,9Z diene pattern, these phospholipid variants did partially incorporate cholesterol. This contrasts with the cases of $\Delta^{5,9}$ -PE(26:2,26:2) and its PC analogue, where considerable broadening of the phospholipid phase-transition peak was observed with added cholesterol (5-30 mol %), thus showing the effects of unsaturation on sterol-fatty acid interactions.

The effect of chain length is more significant. Synthetic phospholipids incorporating the 5Z,9Z diene system in C_{18} – C_{24} acyl groups, namely, $\Delta^{5,9}$ -PC-(18:2,18:2), $\Delta^{5,9}$ -PC(22:2,22:2), and $\Delta^{5,9}$ -PC(24:2,24:2), ⁸⁰ all were found to integrate cholesterol readily into the liposome bilayers, underscoring for the first time the critical effect of the C₂₆ chain length of demospongic

The thermotropic properties of several monoenoic C_{24} -acyl phospholipids were also examined.⁸¹ A chain length of below C_{26} was selected to display only the effect of double-bond location. Both Δ^5 -PC(24:1,24:1) and Δ^9 -PC(24:1,24:1) behaved like $\Delta^{6,9}$ -PC(26:2,26:2), with partial incorporation of cholesterol, whereas in the case of Δ^{15} -PC(24:1,24:1), cholesterol interacted with the saturated acyl regions just as with PC(16:0,16:0). Judging from these preliminary results, both the C₂₆ chain length and the 5Z,9Z diene pattern of demospongic acids were critical to the exclusion of sterols from the sponge phospholipid bilayer.

In our most recent model membrane study, 82 a number of novel marine sterols were found to interact readily with the conventional phospholipids PC-(16:0,16:0) and PC $(18:0,\Delta^9-18:1)$. These phospholipids also interact with the demospongic type $\Delta^{5,9}$ -PC-(26:2,26:2). Do sterols only integrate the short-chain fatty acids in sponge membranes? Is the interaction between demospongic acids and sponge sterols mediated by the short-chain fatty acids?

Evolution in the oxygen-rich atmosphere has dictated the role of sterols in adjusting the thickness, cohesive strength, and fluidity of eukaryotic membranes.83 In prokaryotes and some lower organisms (e.g., Tetrahymenia), the place of sterols is taken by analogous molecules 12,13 or their phylogenetic precursors.84 The

⁽⁷⁹⁾ Li, H.; Düzgünes, N.; Ayanoglu, E.; Djerassi, C. Chem. Phys. Lipids 1988, 48, 109-117.

⁽⁸⁰⁾ Ayanoglu, E.; Li, H.; Djerassi, C.; Düzgünes, N. Chem. Phys. Lipids 1988, 47, 165-175.

⁽⁸¹⁾ Ayanoglu, E.; Chiche, B. H.; Beatty, M. F.; Djerassi, C. Biochemistry 1990, 29, 3466-3471.

⁽⁸²⁾ Dai, M. C.; Chiche, B. H.; Ayanoglu, E.; Düzgünes, N.; Djerassi, Chem. Phys. Lipids, submitted for publication.
(83) Bloom, M.; Mouritsen, O. G. Can. J. Chem. 1988, 66, 706-712 and

literature cited therein. (84) Krajewski-Bertrand, M.-A.; Hayer, M.; Wolff, G.; Milon, A.; Albrecht, A. M.; Heissler, D.; Nakatani, Y.; Ourisson, G. Tetrahedron 1990, 46, 3143-3154.

(1)

existence of novel sponge sterols that do not interact with demospongic phospholipids strongly implicates some other membrane molecules participating in lipid interaction, the most likely candidates being the membrane proteins. 12,13,83,84

We recently identified⁸⁵ a number of conventional (mainly C₁₆-C₁₈) fatty acids covalently linked to intrinsic proteins of marine sponge membranes. Is it probable that these lipoproteins are interacting with the sterols or demospongic acids? In view of the relatively rare occurrence of acylated membrane proteins and their obscure biological roles, 12,86 new insights into these molecules may well be gained from further studies with sponge proteins.

(85) Ayanoglu, E.; Rizzolio, M.; Beaulieu, S.; Roberts, J.; Oz, O.; Djerassi, C. Comp. Biochem. Physiol. 1990, 96B, 597-603. (86) Wilcox, C. A.; Olson, E. N. Biochemistry 1987, 26, 1029-1036 and

literature cited therein.

Concluding Remarks

The sponge cell membrane is unique in terms of its lipid diversity and lack of interaction of its phospholipids with sterols. The biological significance of these characteristics is unknown. They may be the result of neutral mutations⁹ or perhaps represent molecular fossils. In any event, the sponge membrane allows a sponge to adapt to the precarious aquatic environment. with its fluctuating temperature, oxygen content, and osmolarity. Further investigation of the components of this distinct membrane will deepen our understanding of cell membrane function in general.

We gratefully acknowledge the participation of the Cancer Research Institute and Department of Pharmaceutical Chemistry, University of California, San Francisco, in the model membrane studies. Financial support at Stanford was provided by National Science Foundation Grant DMB 86-06249 and by National Institutes of Health Grants GM 06840 and AM 04257.

Controlling Electrochemical Catalysis with Surfactant Microstructures

JAMES F. RUSLING

Department of Chemistry (U-60), University of Connecticut, Storrs, Connecticut 06269-3060 Received July 12, 1990 (Revised Manuscript Received January 3, 1991)

Surfactants are amphiphilic ions or molecules with charged or polar head groups and long hydrocarbon tails. Two of their properties are useful in electrochemistry: adsorption at interfaces and aggregation into supramolecular structures. In the early part of this century, adsorption of surface-active compounds on Hg electrodes was found to suppress unwanted convection. Surfactants became widely used for this purpose. Micelles (Figure 1) are examples of surfactant aggregates. They are formed from soluble surfactants above a critical micelle concentration (CMC). Solutes can bind at the micelle-water interface (Stern layer) and in hydrophobic regions just below this interface.² In 1952, Proske was the first to use micelles to solubilize nonpolar organic compounds in water for electrochemical measurements.3

The past 10-15 years saw many new applications of surfactants in electrochemistry. For example, ion radicals produced at electrodes were stabilized by coulombic and hydrophobic interactions with micelles.³ Reducible and oxidizable probes began to be used to

James F. Rusling was born in Philadelphia, PA, in 1946. He received his B.S. in Chemistry from Drexel University in 1969. After several years as an analytical chemist at Sadtler Research Labs and Wyeth Pharmaceutical Co., he earned his Ph.D. from Clarkson University in 1979. He joined the chemistry faculty at the University of Connecticut in 1979, where he is now Professor of Analytical Chemistry. In 1986 he was Visiting Professor at the University of Southampton, England. His research interests include electrochemistry in organized media, bioelectrochemistry, electrochemical solutions to environmental problems, computer modeling of chemical data, photoelectrocatalysis, and surface spectroelectrochemistry.

Scheme I E° (at electrode)

$$ArX + Q \stackrel{k_1}{\rightleftharpoons} ArX^{\bullet -} + P$$
 (2)

$$ArX^{\bullet -} \xrightarrow{k} Ar^{\bullet} + X^{-}$$
 (3)

$$Ar^{\bullet} + Q \longrightarrow P + Ar^{-}$$
 (fast) (4)

$$Ar^- + (H^+) \longrightarrow ArH$$
 (fast) (5)

measure diffusion in surfactant media.^{4,5} Microstructures of amphiphiles adsorbed on electrodes were investigated. 6 Coatings of functional amphiphiles were prepared on electrodes by Langmuir-Blodgett (LB) methods.7

(1) Heyrovsky, J.; Kuta, J. Principles of Polarography; Academic: New York, 1966.

(2) Fendler, J. H. Membrane Mimetic Chemistry; Wiley: New York,

(3) Proske, G. E. O. Anal. Chem. 1952, 24, 1834-1837.

(4) McIntyre, G. Crit. Rev. Anal. Chem. 1990, 21, 257-278. (5) (a) Shinozuka, N.; Hayano, S. In Solution Chemistry of Surfac-

(5) (a) Shinozuka, N.; Hayano, S. In Solution Chemistry of Surfactants; Mitall, K. L., Ed.; Plenum: New York, 1979; Vol. 2, pp 599-623. (b) Mackay, R. A.; Dixit, N. S.; Agerwal, R.; Seiders, R. P. J. Dispersion Sci. Technol. 1983, 4, 397-407 and references therein.

(6) (a) Mousty, C.; Pouillen, P.; Marte, A.-M.; Mousset, G. J. Colloid Interface Sci. 1986, 113, 521-529. (b) Van Galen, D. A.; Majda, M. Anal. Chem. 1988, 60, 1549-1553. (c) Facci, J. S. Langmuir 1987, 3, 525-530. (d) Diaz, A.; Kaifer, A. E. J. Electroanal. Chem. 1988, 249, 333-336. (e) Donohue, J. J.; Buttry, D. A. Langmuir 1989, 5, 671-678. (f) Widrig, C. A.; Majda, M. Langmuir 1989, 5, 689-695. (g) Bunding Lee, K. A.; Mowry, R.; McLennan, G.; Finklea, H. O. J. Electroanal. Chem. 1988, 246, 217-224. (h) Bunding Lee, K. A. Langmuir 1990, 6, 709-712. (i) Creager, S. E.; Collard, D. M.; Fox, M. A. Langmuir 1990, 6, 1617-1620.